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Introduction K_a K_b
 K_w pH pOH pKa
pKb H^+ OH^-
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(Part 1) - Acid,

Bases And Salts |

Class 10 Chemistry

pH, pOH, H_3O^+ ,

OH^- , K_w , K_a , K_b ,

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all acids and all
bases have in
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Chapter 5 - Acids,
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Bases and Salts L1
| Introduction |
CBSE Class 10

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Chemistry | Don't
Memorise
Calculating pH,
pOH, [H⁺],
[H₃O⁺], [OH⁻] of
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Practice Acids and
Bases, pH and pOH
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How Strong acid
base solution pH
Scale Class 11
chapter 7 |
Equilibrium | Ionic~~

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Equilibrium 01 |

Theories Of Acids
Chapter Test A
and Bases JEE

Answers
MAINS/NEET How
~~Strong are Acid or
Base Solutions? (pH
Values)~~ Acids

Bases And
Solutions Chapter
In this chapter,
students get hold of
basic knowledge on
Acids bases and
salts. In this

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Chapter, various
chemical properties
of acids, bases and
salts and the

reaction of them
with metals, non-
metals are studied.

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10 Chapter 2

Science Acids
bases and salts.

There are a lot of
practical based

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questions in this
Solution and these
types of questions
are often asked in
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science
examination. You
will encounter
questions on a day
to day examples of
...

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Chapter 2 Acid
Bases and ...
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Acids, Bases, and
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(continued) 28. Dip red litmus paper into each solution and watch for color changes. Then do the same with blue litmus paper. Any solution that turns red litmus blue is

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basic. Any solution that turns blue litmus red is acidic.

A solution that does not change the color of either litmus paper is neutral. 29. When acids dissolve in water, they produce hydrogen ions (H^+). Bases

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Solutions
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Solutions
Calculating a
Concentration To
calculate the
concentration of a
solution, compare
the amount of
solute to the
amount of solution
and multiply by 100

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And Solutions
percent. For
example, if a
solution contains 10
grams of solute
dissolved in 100
grams of solution,
then its
concentration can
be reported as 10
percent.

Chapter 7 Acids,
Bases, and
Solutions

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Solutions. study.

play. a well mixed
mixture that

contains a solvent
and at least one
solute. solution.

what is an example
of a solution? salt
water. the part of a
solution present in
the large amounts
(does the
dissolving) solvent.

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SCIENCE TEST
CHAPTER 7 ACIDS
BASES AND
SOLUTIONS ...

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Chapter 5: In this
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Solutions Class 7
Science Chapter 5 –
Acids, Bases And
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provided here are

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Chapter 5 MCQ

(Multiple Choice
Questions) of

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Important questions from chapter 5 of grade 7 science are given here in the form of MCQs for the preparation of class assessments, school tests and terminal exams.

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Class 10 Chemistry
Chapter 2 Important
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Important Questions
Very Short Answer

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Type. Question 1.

How is the concentration of hydronium (H_3O^+) ions affected when a solution of an acid ...

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Important topics covered in NCERT

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Solutions for class
7 chapter 5 Acids
bases and salts The
combination of acid
and base results in
the formation of
salts which can
either be acidic,
basic or neutral in
nature. This
process is known as
neutralisation.
Neutralization is a
very important

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topic with respect
to examination.
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Chapter 2 Acids and

Bases ...

Question 1: (a)

Chloride, nitrate,
hydride, ammonium.

(b) Hydrogen

chloride, sodium
hydroxide, calcium

oxide, ammonia. (c)

Acetic acid,

carbonic acid,

hydrochloric acid,

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nitric acid. (d)
Ammonium
chloride, sodium
chloride, potassium
nitrate, sodium
sulphate. (e)
Sodium nitrate,
sodium carbonate,
...

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Technology
Solutions for Class
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Acids Bases

Acids react with bases to produce a salt compound and water. When equal moles of an acid and a base are combined, the acid is neutralized by the base. The products of this reaction are an ionic compound, which is labeled as a salt, and water.

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Properties of Acids
Chapter Test A
and Bases |

Answers
Chemistry for Non-
Majors

Chapter - 2, Acids
Bases and Salts

Notes. • These are
the substances
which have sour in
taste. • They turns
blue litmus solution
to red. • These
substances are

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bitter in taste. •
They turns red
litmus solution to
blue. • They give
OH ions in aqueous
solution. There are
the substance that
changes their
colour / smell in
different types of
substances.

Chapter - 2, Acids
Bases and Salts

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emilydonels.

Chapter 6 Solutions,
Acids, and Bases.

dilute solution.

base. Hydrogen Ion.

neutralization. a

mixture that has

only a little solute

dissolved in a

certain.... A

substance that

tastes bitter, feels

slippery, and turns

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red... makes
something an acid;
a positively charged
ion (H^+) formed....

acids acids chapter
6 bases solutions
Flashcards and
Study ...

Acids: Acids are
sour in taste, turn
blue litmus red, and
dissolve in water to
release H^+ ions.

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Acids Bases

Example: Sulphuric acid (H_2SO_4), Acetic Acid (CH_3COOH), Nitric Acid (HNO_3) etc.

Properties of Acids:

Acids have a sour taste. Turns blue litmus red. Acid solution conducts electricity. Release H^+ ions in aqueous solution.; Types of Acids: Acids are

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