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Main Points To Be Used In Answering A Past Paper Question:
Down the group there is a change in structure from giant molecular to giant metallic and a change in bonding from covalent to metallic.

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From C to Ge elements exhibit a giant molecular structure. Sn and Pb exhibit a giant metallic lattice structure.

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FUNDAMENTALS IN CHEMISTRY 1. (a)(i) Define the term

bond energy [1. mark] The bond energy represents: The quantity of energy evolved when the bond is formed. Also, the bond energy is the amount of energy absorbed when the bond is broken. (ii) The bond lengths and bond energies of carbon-carbon single, double and triple bonds are listed below.

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