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1. Gases consist of tiny particles (atoms or

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molecules) 2. These particles are so small, compared with the distance between them, that the volume (size) of the individual particles can be assumed to be negligible (zero) 3. The particles

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A gas is a state of matter with no defined shape or volume. Gases have their own unique behavior depending on a variety of variables, such as temperature, pressure, and volume. While each gas is different, all gases act in a similar manner. This study guide highlights

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the concepts and laws dealing with the chemistry of gases.

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STUDY GUIDE AP Chemistry Chapter Ten- Gases

Lecture Notes 101 Characteristics of Gases

- All substances have three phases: solid, liquid and gas
- Substances that are liquids or solids under ordinary conditions may also exist as gases
- These are often referred to as vapors
- Many of the properties of gases differ

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Lecture Notes 10.1 Characteristics of Gases

- All substances have three phases: solid, liquid and gas.
- Substances that are liquids or solids under ordinary conditions may also exist as gases.
- These are often referred to as vapors.
- Many of the properties of gases

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