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different compounds. In one compound, 0.853 g of fluorine combines with 1.472 g of xenon. In the other compound, 0.624 g of fluorine combines with 2.16 g of xenon. Do these data support the law of multiple proportions? Show your work ...

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EXAMPLE PROBLEM 11-1 For more practice converting from representative particles to moles, go to Supplemental Practice Problems in Appendix A. a c t i c e !

Ointments containing zinc oxide provide protection from sunburn and are used to treat some skin diseases. Section 11.1

Assessment 5. How is a mole similar to a

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CHAPTER Practice Problems 11.1 The Many Forms of Energy pages 285 – 292  
page 287 1. A skater with a mass of 52.0 kg moving at 2.5 m/s glides to a stop over a distance of 24.0 m. How much work did the friction of the ice do to bring the skater to a stop?



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CHAPTER 11 Energy and Its Conservation

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Identify and calculate the number of representative particles in each of the following quantities. a. 2.15 moles of gold. b. 0.151 mole of nitrogen oxide . c. 11.5 moles of potassium bromide . 2. Calculate the

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number of moles of the substance that  
contains the following number of  
representative particles.

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Supplemental Problems Additional  
Challenge Problems Pre-AP/Critical  
Thinking Problems Physics Test Prep:  
Studying for the ... and Challenge Problems  
for each chapter, as well as the Additional  
Problems that appear in Appendix B ... 11. a.  
139 cm 2.3 cm 320 cm<sup>2</sup> or 3.2 10<sup>2</sup> cm<sup>2</sup> b.  
3.2145 km 4.23 km 13.6 km<sup>2</sup> 12. a. 13.78 g

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that you require to acquire those every needs  
considering having significantly cash?

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Problem 4 Godwit Associates paid \$60,000  
for a 20-seat skybox at Memorial Stadium

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for eight professional football games.

Regular seats to these games range from \$70 to \$150 each. At one game, an employee of Godwit entertained 18 clients. Godwit furnished food and beverages for the event at a cost of \$950.

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Matter and Change • Chapter 11 209  
Stoichiometry Stoichiometry CHAPTER 11  
SOLUTIONS MANUAL Section 11.1  
Defining Stoichiometry pages 368 – 372  
Practice Problems pages 371 – 372 1.  
Interpret the following balanced chemical

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equations in terms of particles, moles, and mass.

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Stoichiometry Answers

Supplemental Problems 8. Determine the molar mass of each of the 9. following compounds. a. formic acid ( $\text{CH}_2\text{O}_2$ ) b.

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ammonium dichromate ( $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ ) 42.27 g  
-zsa . What is the mass in grams of each of  
the following quantities?  
3 a. 2.53 moles  
( $\text{Pb}(\text{NO}_3)_2$ ) 32 b. 4.62 moles of magnesium  
bromide ( $\text{MgBr}_2$ ) Calculate the number of  
moles in each of the 10. 11.

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improved and faster can be undergone by  
some ways. Experiencing, listening to the  
new experience, adventuring, studying,

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training, and more practical comings and goings may back you to improve. Chapter 11 Supplemental Problems Stoichiometry Answers Chapter 11 Supplemental Problems Answers

Chapter 11 Supplemental Problems Answers  
Chapter 8 Supplemental Problems 1. Which

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of the following properties from an income tax standpoint will be the best one for Jaun to contribute to his favorite charity? a. Stock acquired in 1992, basis = \$13,000, FMV = \$10,000.



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