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25: Section 1:

Nuclear

Radiation

Chapter 25

Lesson 25.1

Nuclear

Radiation-

Chemistry by

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Transformation

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1 of 9 Nuclear
Chemistry: Crash
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~~#38 Chemistry 1
Notes - Ch 25
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~~Nuclear
Decay Chapter 25
- Applications
of Radioactivity~~

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~~Nuclear Half
life Simulation~~

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*radiation? How
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*how it works |
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Unstable Nuclei

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Chemistry Ch 25

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Terms in this
set (37) Neutron
Ejection. when a
neutron is

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emitted from the
nucleus. ${}^1_0\text{n}$.

Particle for
Neutron

Ejection. ${}^5_2\text{He} \rightarrow$
 ${}^1_0\text{n} + {}^4_2\text{He}$.

Example of
Neutron
Ejection.

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central. Key

Concepts: Terms
in this set
(103) beta

particle. a fast-moving electron formed by the decomposition of a neutron. half-life.

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Discovers

radiation.

Pierre and Marie

curie. Proposed

the definitions

for radiation

and

radioactivity.

Nuclear

reactions.

Reactions that

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Nuclear
involve
radioisotopes
and are not
affected by
temperature
pressure or a
catalyst.
Chemical
reactions.

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chapter 25

nuclear

chemistry study

In this chapter you will study nuclear chemistry, which is concerned with the structure of atomic nuclei and the changes they undergo. An

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application of a nuclear reaction is shown in the photo of the human neck and skull.

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Types of

Radiation

Page 23/45

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Discuss Explain
that the nuclei
of a radioactive
element

spontaneously
decompose.

Nuclear
chemistry is the
study of changes
in matter that
originate in
atomic nuclei.

Ask, What types
of radi-ation

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exist, and how harmful are they? (The three most common types of radiation emitted by unstable nuclei are

~~25.1 Nuclear Radiation~~
25
Chemistry
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Nuclear

Chemistry

Notes/Questions

??questionNucleon

answerproton +

neutron

questionIn

nuclear

chemistry, an

atom is referred

to as a what?

what is it

identified by?

answernuclide;

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by the

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Chemistry:

Matter and

Change Chapter

25. Name Class

Section 25.4

Fassñon and

Page 27/45

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ft5sion of
Atomic Nuclei In
your textbook,
read about the
process of by
which electrical
energy is
produced in a
nuclear power
plant. Use the
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diagram to
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terms, and more
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CHAPTER 25 What
You'll Learn You
will trace the
history of
nuclear
chemistry from
dis-covery to
application. You
will identify
types of
radioactive

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decay and solve
decay rate
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fossil fuel
power plant, the
chemical
combustion of
coal, oil, or
gas generates
the heat whereas
in a nuclear
power plant, a
nuclear fission
generates heat.
The binding
together of two
or more light

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(less than 60
mass number) and
less stable
nuclei to form a
single more
stable nucleus

~~Quia~~

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Section Goals
and

Introductions

Section 18.1 The
Nucleus and

Radioactivity

Goals To

introduce the
new terms

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nucleon, nucleon number, and nuclide. To show the symbolism used to represent nuclides. To explain why some nuclei are stable and others not. To provide you with a way of predicting

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nuclear
stability.

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will trace the
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chemistry from
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will identify
types of
radioactive
decay and solve
decay rate
problems. You
will describe
the reac-tions
involved in
nuclear fission
and fusion. You

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will learn about
applications of
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reactions

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10 Chapter 4

Characteristics
of Waves.47 10.

What is the
highest atomic
number shown on
the periodic
table? 10 Guided
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Fission and

fusion of

Atomic Nuclei In

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your textbook,
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nuclear power
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