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6.1 Introduction to Chemical Bonding FSc Chemistry Book1,  
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Atom IIT JEE NEET FSC CHEMISTRY BOOK 1 CH 6- MCQS  
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LEC 9: Electron Affinity Ch 6 Lec 3 Atomic Radii. Chemical

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CHAPTER 6 TEST Chemical Bonding MULTIPLE CHOICE

On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question. 1. An ionic bond results from electrical attraction between large numbers of . a. cations and anions c. dipoles b. atoms d. orbitals 2. A nonpolar covalent bond is unlikely when two different atoms join ...

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~~Name ...~~

Chemists often theorized that a molecule that contains a single bond and a double bond split its time existing as one

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of these two structures. This effect became known as \_\_\_\_\_ . a. alternation

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Chapter 6 PRETEST: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

1. The charge on an ion is a. always positive.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

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White. chemical bond: a mutual electrical attraction

between the nuclei and valence electrons of different atoms

that binds the atoms together: ionic bonding:

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Chapter 6 Vocabulary: Chemical Bonding, Chapter 6 Chem

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District CHAPTER 6 REVIEW Chemical Bonding SECTION 1

SHORT ANSWER Answer the following questions in the

space provided. 1. a A chemical bond between atoms results

from the attraction between the valence electrons and of

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different atoms.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

## ~~Chapter 6 Chemical Bonding Test Answer Key~~

CHAPTER 6 TEST: CHEMICAL BONDING REVIEW SHEET. 1. What are the 2 main types of bonds that we have learned about? ... What levels of differences in electronegativity correspond to polar covalent, nonpolar covalent, and ionic bonds? 6. If Sodium, Na, was to bond with Chlorine, Cl, what

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type of bond would they form? ...

~~CHAPTER 6 TEST: CHEMICAL BONDING REVIEW SHEET~~

Chapter 6 PRETEST: Chemical Bonding Accelerated  
Chemistry 3 Chemical Bonding Chapter 6 Pre-test,  
continued 11Two atoms will likely form a polar covalent  
bond if the electronegativity difference is a 01 b 10 c 25 d 40  
12In which of these compounds is the bond between the  
atoms not a Chapter 6 Chemical Bonds Physical Science  
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