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Chemistry Chapter 12 Solutions Vocabulary. a  
homogeneous mixture of 2 or more substances uniformly  
dispersed throughout a single phase. in a solution, the  
substance in which the solute dissolves. the diffusion of  
water or another solvent from a more dilute solution (of a  
solute) to a more concentrated solution (of the solute)  
through a membrane that is permeable to the solvent.

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the physical state in which the opposing process of dissolution and crystallization of a solute occur at equal rates. Saturated solution. a solution that contains the maximum amount of dissolved solute. Unsaturated solution. a solution that contains less solute than a saturated solution under the existing conditions.

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Chemistry | Chapter 12 | Solutions. the ability of one substance to dissolve in another at a given temperature and pressure; expressed in terms of the amount of solute that will dissolve in a given amount of solvent to produce a saturated solution. the law that states that at constant temperature, the solubility of a gas in a liquid is directly proportional to the partial pressure of the gas on the surface of the liquid.

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Important Questions for Class 12 Chemistry Solutions

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soluble. capable of being dissolved. solution. a

homogeneous mixture of two or more substances in a single

phase. solvent. the dissolving medium in a solution. solute.

the substance dissolved in a solution. suspension. Chemistry

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Flashcards | Quizlet Chapter 12: Solutions. Lessons. List and explain 3 factors that affect the rate at which a solid solute dissolves in a liquid solvent.

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