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sphere of homogeneous solution. 6.2 EXPLAINING SOLUTIONS PRACTICE (Page 273) Understanding Concepts 1. Most molecules are polar. The four categories of polar molecules are: AB (e.g., HCl (g), CO (g)) $N \times A \times y$ (e.g., NH₃(g), NF₃(g)) $O \times A \times y$ (e.g., H₂O (l), OCl₂(g)) $C \times A \times y \times B \times z$ (e.g., CHCl₃(l), C₂H₅OH (l)) 2.

6.1 DEFINING A SOLUTION

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