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Cathodic Protection Four parts needed for a CP cell to exist 1. Anode □Where corrosion occurs 2. Cathode □Protected from corrosion 3. Electrolyte □Soil or water (any conductive environment) adjacent to □and containing both the anode and the cathode 4. Metallic Path □Physically connects the

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Corrosion.ppt 1. PREPARED BY: DARSHAN R. KANJIYA 11MCY 043 KSKV KACHCHH UNIVERSITY, DEPARTMENT OF CHEMISTRY, BHUJ 1 2. 3. INTRODUCTION Corrosion is a process of formation of the compound of pure metal by the chemical reaction between metallic surface and its environment. It is an oxidation process.

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protective Measures Against Corrosion Organic Coating : Apply on metallic surface for protection from corrosion & also to impart decorative value. Such as paints, enamel, Varnishes & lacquers Corrosion Inhibitors Substances which when added in a small amount in an environment reduces the rate of corrosion of a metal exposed to that environment.

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corrosion protection for a wide variety of environmental conditions. Nevertheless, the choice of the material or the corrosion protection ... We only talk about corrosion when there is a change in the metal's or system's properties which may lead to an undesirable outcome. This can range simply from

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Our Guides to Good Practice in Corrosion Control provide advice concerning corrosion control, including corrosion protection schemes and corrosion management for a variety of specific applications in the construction industry.

Corrosion good practice guides - NPL

This chapter presents important aspects of corrosion in industrial infrastructure, its causes, impacts, control, protection and prevention methods. 2. Materials in industry Metallic materials play a key role in the development of a country and its sustained growth in the context of the global economy.

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