

## Counting Atoms Sheet Answers

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Counting Atoms

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The Mole: Counting Atoms **Atomic Number, Atomic Mass, and the Atomic Structure | How to Pass Chemistry**

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of atoms of each element or polyatomic ions is written in the space  
below the line and to the right of the symbol as a subscript. Six  
times and count them all. 2 carbon c has a subscript of 1. Designed

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as drill practice to reinforce. Al 2 so 4 3 4na 2co 3. None means 1  
1. A subscript is a number written at the lower right corner behind  
the symbol of an element. Try these ones count the number of atoms of  
each element in ...

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Counting Atoms & Diagramming Chemical Equations Worksheet Part A:

Directions: 1) If there is a subscript, circle it. 2) If there is a  
coefficient, draw a square around it. 3) Count how many atoms of each  
element there are. HINT: If there is no coefficient AND no subscript  
right after the chemical symbol, there is 1 atom of that element.

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Element # of Atoms Element # of Atoms Element # of Atoms Ca 3 Fe 2 Al  
2 P 2 C 3 S 3 O 8 O 9 O 9 Total 13 Total 14 Total 14 ANSWER KEY  
element symbols subscripts right atoms one subscript right brackets  
multiplied. Created by Anh-Thi Tang – Tangstar ...

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2KN03 9MgS04 Sr = Ca = Al = Au = 3D 30 n (Continued) INSTRUCTIONS:  
Write the quantity of atoms of element opposite the formula of the  
compound for the quantity of formula units and molecules shown: ) 24  
For example: 5Pz03 For 4Zn(NO 3/2 (5 x 2 = ) 10 32. D 20 Mn = -(4x1 x  
So Inc) 19.

**NAME: BLOCK: DATE: CHEMISTRY: COUNTING ATOMS IN COMPOUNDS ...**

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How to Count Atoms Worksheet 1. The symbol of an element represents one atom of that element. e.g., Ba = 2. A subscript is a number written at the lower right corner behind the symbol of an element. If there is more than one atom of the element, then a subscript is used to indicate the number of.

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