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An Introduction to Thermal Physics by Daniel V. Schroeder
Summary This text provides a balanced, well-organized treatment of thermodynamics and statistical mechanics, making thermal physics interesting and accessible to anyone who has completed a year of calculus-based introductory physics. Solutions Schroeder Thermal Physics

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Recommended problems

An Introduction to Thermal Physics

1. DIESEL ENGINES 2. As in the Otto cycle, the red and blue
curves are adiabats, so no heat is exchanged during these steps. The
efficiency of the Diesel cycle is most easily worked out by

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calculating the heat input Q along the yellow step 2 to 3 and the heat expelled during the green step 4 to 1. The efficiency is then $e = W/Q$.

DIESEL ENGINES An Introduction to Thermal Physics
in Schroeder), $\Delta H_{H_2O} = 285.83 \text{ kJ mol}^{-1}$ and $\Delta H_{H_2} = 890.36 \text{ kJ mol}^{-1}$ (12) This time, the final volume is $1/3$ of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work $P\Delta V = RT\Delta n$ (13) $= 8.31 \text{ J K}^{-1} (298 \text{ K})(2)$ (14) $= 4.953 \text{ kJ}$ (15)

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