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Name SECTION 3 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L.

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~~Chapter 11 - States of Matter and Intermolecular Forces~~

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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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Chemistry: Chapter 3 Section 3. terms to know for Chapter 3 Section 3 in Chemistry. STUDY. PLAY. Atomic Number . number of protons in the nucleus of each atom in that element. Isotopes. atoms of the same element that have different masses. Mass Number. the total number of protons and neutrons in the nucleus of an isotope. Nuclide. a general term for any isotope of any element. Atomic Mass Unit ...

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