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SECTION 3 SHORT ANSWER Answer the following

questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope. [http://mchsapchemistry.com/uploads/7/7/1/5/7715124/ch\\_3\\_study\\_guide\\_key.p...](http://mchsapchemistry.com/uploads/7/7/1/5/7715124/ch_3_study_guide_key.p...)

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Indicate whether each of the following is a physical or  
chemical property of water.

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom's

## 5 The Periodic Law

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