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element with an even atomic number is more abundant than both elements with the adjacently larger and smaller odd atomic numbers. This tendency of the abundance of the chemical elements was first reported by Giuseppe Oddo in 1914 and William Draper Harkins in 1917. Estimated abundances of the chemical

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Abundance is greater than that of adjacent elements with an odd atomic number. Consequently, a graph plotting relative atomic abundance against increasing atomic number ( $Z$ ) displays a 'toothed' curve, rather than a smooth line.

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Oddo—Harkins rule | Encyclopedia.com

Oddo – Harkins rule. Source: A Dictionary of Earth Sciences. Author (s): MICHAEL ALLABYMICHAEL ALLABY. Rule stating that the cosmic abundance of elements with an even atomic number is greater than that of adjacent elements ... Access to the

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The Oddo-Harkins rule holds that elements with an even atomic number (such as carbon) are more common than elements with an odd atomic number

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(such as nitrogen). The effect was first reported by Giuseppe Oddo [1] in 1914 and William Draper Harkins [2] in 1917. [3]

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Oddo-Harkins rule of element abundances

The Oddo-Harkins rule holds that elements with an even atomic number are more common than elements with an odd atomic number. The effect was first reported by Giuseppe Oddo in 1914 and William Draper Harkins in 1917.



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the Oddo-Harkins rule (odd atomic  
number elements tend to be Oddo-  
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Oddo-Harkins rule holds that elements  
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common than elements with an odd atomic number. The effect was first reported by Giuseppe Oddo in 1914 and William Draper Harkins in 1917. What does Oddo-Harkins rule mean?

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that meant to introduce students to the concept of isotope ratios, simple counting statistics, intrinsic instrument bias, correlated errors, analytical precision, and analytical accuracy

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When scientists started discovering the elements, they noticed a strange pattern. Today, that pattern is known as the Oddo-Harkins Rule. And while we understand some of the reason for the rule,...

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Oddo-Harkins rule describes about the undulatory behavior of even-odd mass



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elements in universe, even mass elements  
are most abundant. What are the possible  
reasons for this unique selection?

Why does the universe like more even  
mass element than Odd?

Published on Apr 22, 2015 Video shows  
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which states that elements that have an even number of protons in the nucleus are more common than those with an odd...

Oddo-Harkins rule Meaning

The Oddo-Harkins rule simply states that the abundance of elements with an even atomic number is greater than the

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Abundance of elements with an odd atomic number. In order to obtain these elements, the minerals must go through a separating process, known as separation chemistry. This can be done with selective reduction or oxidation.

Lanthanides: Properties and Reactions -

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The Oddo – Harkins rule holds that elements with an even atomic number (such as carbon) are more common than elements with an odd atomic number (such as nitrogen). This effect on the abundance of the chemical elements was first reported by Giuseppe Oddo in 1914

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and William Draper Harkins in 1917.

Translation of Oddo-Harkins rule in  
English

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elements with an even atomic number  
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elements with an odd atomic number

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(such as nitrogen). This effect on the abundance of the chemical elements was first reported by Giuseppe Oddo in 1914 and William Draper Harkins in 1917.

Oddo – Harkins rule — Wikipedia  
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Geochemically, therefore, they are

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lithophile refractory elements. Given that the chemical behavior of the REE is so similar, their abundance patterns in geological materials illustrate well the Oddo-Harkins Rule, i.e., the abundance of Z- odd elements is smaller than the abundance of neighboring Z-even elements.

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Geochemistry of the Lanthanide Elements

Oddo-Harkins rule. Oddo-Harkins rule: translation. noun. a rule which states that elements that have an even number of protons in the nucleus are more common than those with an odd number.



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